TC1700 Bldg./R.R.H.MSEN

U. S. DEPARTMENT OF COMMERCE PATENT AND TRADEMARK OFFICE

Organization_

IF UNDELIVERABLE RETURN IN TEN DAYS WASHINGTON, DC 20231

OFFICIAL BUSINESS

AN EQUAL OPPORTUNITY EMPLOYER

Aven Cf Otten 11 Promona

REVICIL OGGOLZGIG 1203 OG 04/27/04 FORWARD TIME EXP RTN TO SEND REVIS'STEPHEN 1 ABBOTTSFORD AVON CT 06001-3953

RETURN TO SENDER

PECEIVED MAY 1.1 2004 ⁷C 7²00

1745



UNITED STATES PATENT AND TRADEMARK OFFICE

UNITED STATES DEPARTMENT OF COMMERCE
United States Patent and Trademark Office
Address: COMMISSIONER FOR PATENTS
P.O. Box 1450
Alexandria, Virginia 22313-1450
www.usplo.gov

APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO	
09/992,591	11/06/2001	Paul R. Margiott	C-2451	7404	
75	7590 05/11/2004		EXAM	EXAMINER	
Stephen E. Revis 1 Abbottsford			ALEJANDRO, RAYMOND		
	001-3953		ART UNIT	PAPER NUMBER	
			1745		
			DATE MAILED: 05/11/2004		

Please find below and/or attached an Office communication concerning this application or proceeding.

		Application No.	Applicant(s)	
		09/992,591	MARGIOTT ET AL.	
	Office Action Summary	Examiner	Art Unit	
		Raymond Alejandro	1745	
Period for I	The MAILING DATE of this communication app	ears on the cover sheet with the c	orrespondence addre	
A SHOF THE MA - Extensic after SIX - If the pe - If NO pe - Failure t Any repl	RTENED STATUTORY PERIOD FOR REPLY ALLING DATE OF THIS COMMUNICATION. In sof time may be available under the provisions of 37 CFR 1.13. (6) MONTHS from the mailing date of this communication. iod for reply specified above is less than thirty (30) days, a reply riod for reply is specified above, the maximum statutory period we be reply within the set or extended period for reply will, by statute, by received by the Office later than three months after the mailing latent term adjustment. See 37 CFR 1.704(b).	86(a). In no event, however, may a reply be time within the statutory minimum of thirty (30) days ill apply and will expire SIX (6) MONTHS from cause the application to become ABANDONEI	nely filed s will be considered timely. the mailing date of this comm D (35 U.S.C. § 133).	nunication.
Status				
2a)⊠ TI 3)⊡ Si	esponsive to communication(s) filed on <u>19 Fe</u> nis action is FINAL . 2b) This nce this application is in condition for allowant posed in accordance with the practice under E	action is non-final. ace except for formal matters, pro		erits is
Disposition	of Claims			
4a 5)□ Cl 6)⊠ Cl 7)□ Cl	aim(s) 1-17 is/are pending in the application. Of the above claim(s) is/are withdraw aim(s) is/are allowed. aim(s) 1-17 is/are rejected. aim(s) is/are objected to. aim(s) are subject to restriction and/or			
Application	Papers			
10)⊠ Th Ar Re	e specification is objected to by the Examiner e drawing(s) filed on <u>06 November 2001</u> is/ar eplicant may not request that any objection to the coplacement drawing sheet(s) including the correction on the coplacement of the correction of the contraction of the contraction of the contraction is objected to by the Examiner.	re: a)⊠ accepted or b)⊡ objected are by accepted or by objected are by ance. See on is required if the drawing(s) is obj	e 37 CFR 1.85(a). ected to. See 37 CFR	1.121(d).
Priority und	ler 35 U.S.C. § 119			
a) <u>□</u> 1. 2. 3.	Certified copies of the priority documents Certified copies of the priority documents	have been received. have been received in Application ty documents have been receive (PCT Rule 17.2(a)).	on No d in this National Sta	age
Attachment(s)		ОП 2	(DTO 440)	
2) Notice of 3) Informati	References Cited (PTO-892) Thraftsperson's Patent Drawing Review (PTO-948) Thraftsperson's Patent Drawing Review (PTO-948) Thraftsperson's Patential Patential Pto-1449 or PTO/SB/08) Thraftsperson Pto-1449 or PTO/SB/08) Thraftsperson Pto-1449 or PTO/SB/08) Thraftsperson Pto-1449 or PTO/SB/08)	4) Interview Summary (Paper No(s)/Mail Da 5) Notice of Informal Pa 6) Other:	te	o2)

DETAILED ACTION

Response to Amendment

This communication is in response to the amendment filed 02/19/04. Applicant's arguments filed along with the foregoing amendment have been entirely examined and scrutinized but they are not convincing. Thus, the present claims are finally rejected as the 35 USC 103 rejection still stands for the reasons of record:

Claim Rejections - 35 USC § 103

- 1. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
 - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 2. This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).
- 3. Claims 1-17 rejected under 35 U.S.C. 103(a) as being unpatentable over Takechi et al 5154986 in view of Cameron 4537839.

Art Unit: 1745

The present application is directed to a procedure for shutting down a fuel cell and fuel processing system.

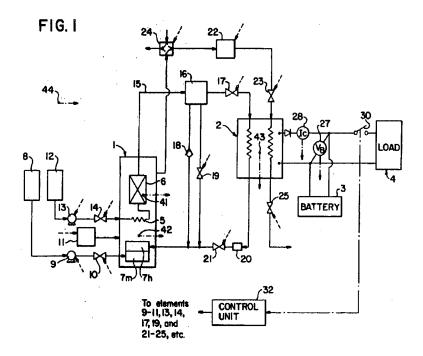
With respect to claims 1 and 15:

Takechi et al disclose shut-off device for fuel cell system wherein in order to achieve a safe shut-down, the fuel cell system must be left in such a state after the cessation of operations that there will be no problem in restarting operations for the fuel cell system (COL 3, lines 59-65). In order for there to be a safe shut-down, the following conditions must be met depending n the type of system to which the control system unit is applied (COL 3, lines 65-68), particularly: condition 3, new air or inert gas beyond certain level must be present in the fuel cell (COL 4, line 3-4), wherein condition 3 indicates that water vapor remains inside the fuel cell. If water vapor remains within the fuel cell in the electrolyte, this water can cause early deterioration of the fuel cell. Safe shut down is accomplished by connecting the four-way valve 2 to the outside air and by using blower 22 to force new air into the cell, or to introduce an inert gas into the cell for a set period of time following the shut-down of the fuel cell (COL 4, lines 26-35). Additionally, instead of eliminating (water migration is reduced) the water vapor by performing the above operation for a set period of time, it is also within the scope of the invention to place a water vapor sensor in the fuel cell and use the detection signal from that sensor to halt the purging operation and ensure a safe shut-down (COL 4, lines 35-41). It is also disclosed that cool outside air is introduced to cool down and ensure a safe shut down (COL 4, lines 47-50). It is disclosed that the operation of the fuel cell is halted while turning off the external load through a main switch (COL 1, lines 29-35).

Art Unit: 1745

Takechi et al disclose that new air or inert gas beyond a certain level must be present in the fuel cell (COL 4, lines 3-4); Takechi et al also describe that safe shutting is accomplished by connecting the four-way valve 24 to the outside air and by using a blower 22 to force new air into the cell, or to introduce an inert gas such as nitrogen into the cell for a set of period of time following the shut-down of the fuel cell (COL 4, lines 29-38). It is further disclosed that instead of eliminating the water vapor, it is also within the scope of the invention to place a water vapor sensor in the fuel cell and use the detection signal from that sensor to halt the purging operation and ensure a safe shut-down (COL 4, lines 35-42).

Figure 1 below illustrates the fuel cell processor 1 for generating hydrogen. Connected in parallel to fuel cell 2 is a storage battery 3. A load 4 is connected to both fuel cell 2 and the battery 3 (COL 2, lines 23-31). A fuel tank 8 storing methanol fuel for the burner 7m where it is burned with the aid of air supplied (COL 2, lines 40-48). 7M and 7H also represent burners (COL 2, lines 37-39).



Art Unit: 1745

Figure 3 below depicts the shut-down method for a fuel cell system:

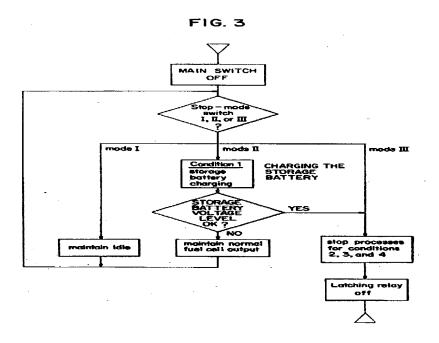
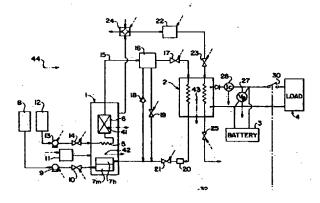
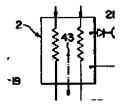


Figure 1 and its enlarged section below shows the fuel cell structure. It is noted that the top and bottom part of feature 2 represents the two fuel cell ends. It is also disclosed heating burners 7M and 7H making hot gases (COL 2, lines 35-45). These burners are in fluid communication with the fuel cell 2. It is also disclosed that valve 24 is connected thereto, allowing the selection of either low temperature atmospheric air or high temperature air from fuel processor 1 (COL 2, lines 62-67).



Art Unit: 1745



Takechi et al is applied, argued and incorporated herein for the reasons above. However,

Takechi et al do not expressly disclose the specific purging steps.

Cameron discloses a fuel cell that may be purged with inert gas and/or preheated and/or maintained at an elevated temperature by combusting the hydrogen fuel with air and passing the combusting product gases either through the cell or through a heat exchanger for imparting heat (ABSTRACT). Preferably the combustor includes a catalyst. The hydrogen can either by taken directly from the fuel supply or the waste hydrogen. If the product gases are to be passed into the cell, the stoichiometry of the combustion should be controlled so that hydrogen is absent (ABSTRACT). Thus, the concentration of hydrogen is below 4 %.

In light of these disclosures, it would have been obvious to one skilled in the art at the time the invention was made to perform the specific purging steps of Cameron in the fuel cell system of Takechi et al because Cameron teaches that his invention provides an inert gas fuel cell purge which is independent of the availability of an external inert gas source; and it provides for the temperature of the fuel cell to be maintained at an uniform temperature. Additionally, Cameron teaches that it has been found that hydrogen fuel may be combusted and the combustion gases passed through the fuel cell to act as the purge. Thus, the combustion product gases are for use either as a fuel cell purge stream or as a heat exchange medium for the fuel cell reactants for heating or maintaining the fuel cell at constant temperature.

Response to Arguments

- 4. Applicant's arguments filed 02/19/04 have been fully considered but they are not persuasive. The main contention of applicants' arguments is premised on the assertion that the prior art fails to teach the step of purging the reformer of residual hydrogen by flowing air through the reformer as well as purging the anode flow field purge. Nevertheless, this assertion is respectfully disagreed with because the primary reference discloses that:
- i) in order for there to <u>be a safe shut-down</u>, the following conditions therefore must be met depending on the type of system to which the control unit is applied: 3. <u>new air or inert gas</u> beyond a certain level <u>must be present in the fuel cell</u> (Takechi et al, COL 4, lines 3-4);
- ii) <u>inert gas</u> beyond a certain level <u>must be present</u> in the catalyst layer of the fuel processor, <u>so that activity in the reaction layer ceases</u> (Takechi et al, COL 4, lines 5-9);

and, iii) that safe shutting is accomplished by connecting the four-way valve 24 to the outside <u>air</u> and by using a blower 22 <u>to force new air into the cell</u>, or to introduce an inert gas such as nitrogen into the cell for a set of period of time <u>following the shut-down of the fuel cell</u> (Takechi et al, COL 4, lines 29-38). It is further disclosed that instead of eliminating the water vapor, it is also within the scope of the invention to place a water vapor sensor in the fuel cell and use the detection signal from that sensor <u>to halt the purging operation and ensure a safe shut-down</u> (Takechi et al, COL 4, lines 35-42);

thus, having shown that the primary references clearly teaches the use of air thereto, it is contended that the air being passed to assist cooling or reducing temperature in the fuel processor (where the reforming process take places), to certain degree, may pick or purge residual hydrogen from the reformer (the fuel processor) because the transport mechanism of the air

Art Unit: 1745

flowing therethrough provides sufficient momentum energy to, at least to some extent, remove or purge any residual material in the fuel processor. That is to say, the examiner verily believes that such cooling air employed therein to reduce temperature necessarily causes the incidental (secondary) effect of purging residual material due to the energy level associated with the transport phenomenon of the flowing air. Unless applicants provide objective evidence demonstrating that such cooling air flowing through the fuel processor does not exhibit sufficient transport energy and/or does not purely behave as expected by fluid mechanic theory, the instant claims are deemed to be obvious over the prior art of record in view of the teachings above promoting the present of air. Furthermore, absent further description of the purging degree or residual removal rate in the present claims, it is additionally asserted that the cooling air passing through the fuel processor, at least, is able to carry out minor amounts of residual hydrogen therefrom, and thus, meet the necessary functional interrelationship to satisfy the claimed purging requirement.

With respect to purging the fuel cell anode flow field, the examiner wishes to point out that since the prior art of record plainly discloses that new air or inert gas beyond a certain level must be present in the fuel cell; or to introduce an inert gas such as air/nitrogen into the cell for a set of period of time (see items i and iii above), the prior art encompasses introducing or circulating air in the fuel cell per se. In that, it is noted that, as well known in the art, the fuel cell is composed of two major reacting (active) parts/sections: a) the cathode side and b) the anode side. Thus, it is further asserted that since the prior art discloses that air must be present in the fuel cell, it (prior art) thus envisions to force or introduce air into the fuel cell parts and ipso facto into both the cathode side and the anode side. In this regard, it is also noted that applicants have

Art Unit: 1745

presented no reasons or objective evidence to support why the prior art (Takechi et al) cannot or may not be able to introduce air in the anode flow field, or (*stated in another way*) introduce air only in the cathode side of the fuel cell.

Applicant's arguments fail to comply with 37 CFR 1.111(b) because they amount to a general allegation that the claims define a patentable invention without specifically pointing out how the language of the claims patentably distinguishes them from the references.

Conclusion

5. THIS ACTION IS MADE FINAL. Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Raymond Alejandro whose telephone number is (571) 272-1282. The examiner can normally be reached on Monday-Thursday (8:00 am - 6:30 pm).

Art Unit: 1745

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Patrick J. Ryan can be reached on (571) 272-1292. The fax phone number for the organization where this application or proceeding is assigned is 703-872-9306.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

Raymond Alejandro Examiner

Art Unit 1745